

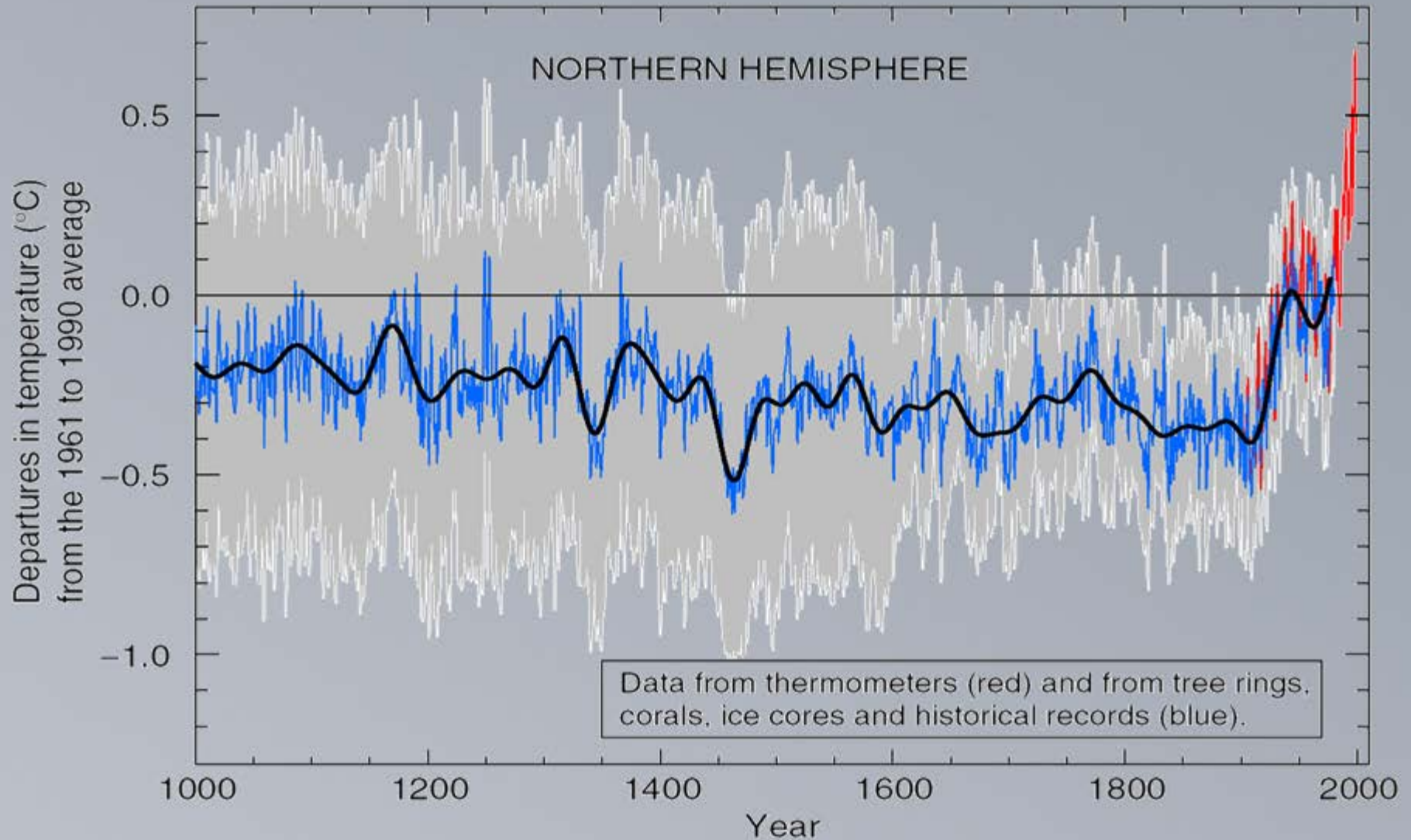
M09. Greenhouse Gas Basics



Source: i-fink.com

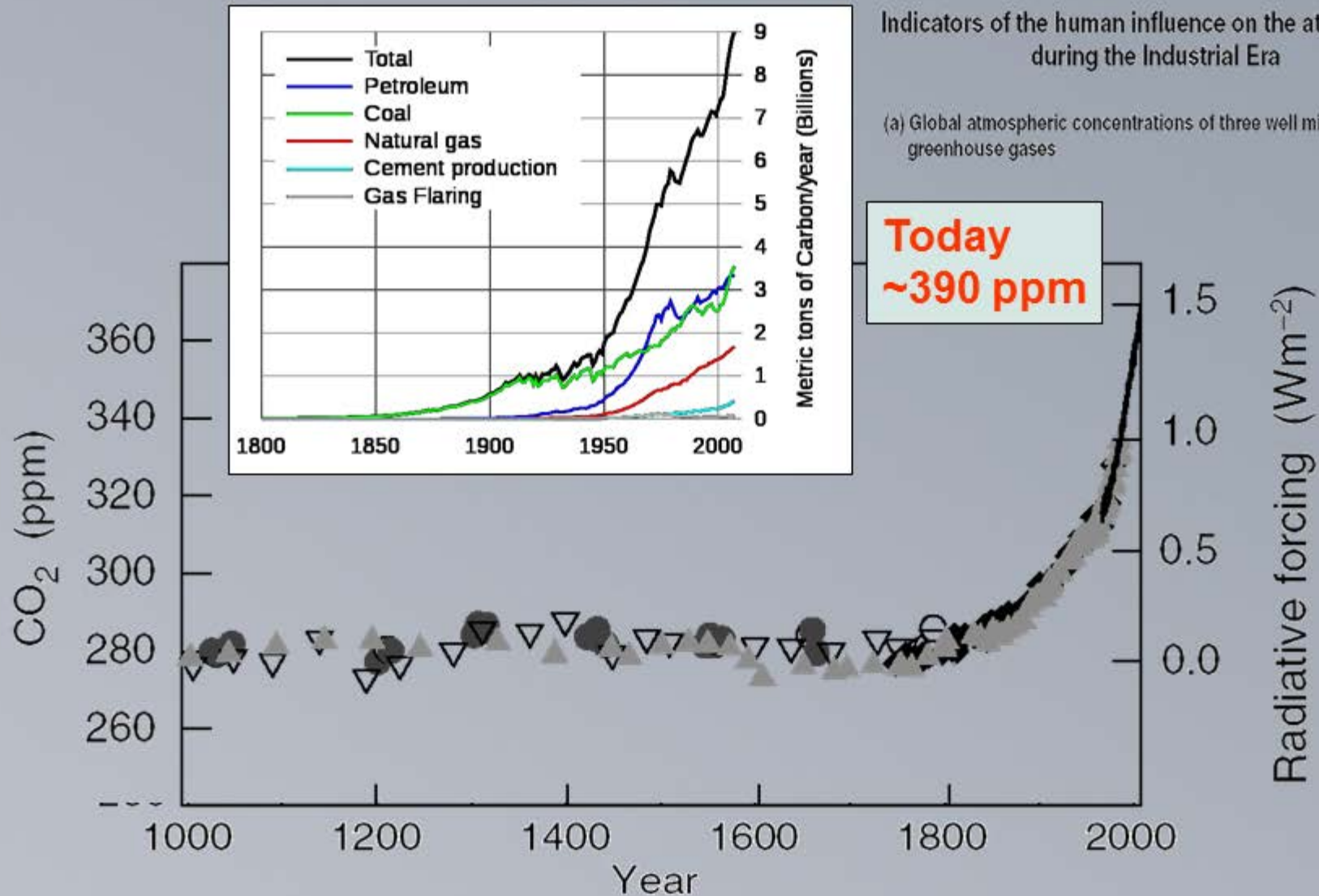
- ✦ Evolution of temperature and CO₂ in air
- ✦ Physical interpretation of global warming
- ✦ Prediction and impacts of climate change
- ✦ Relevant greenhouse gases (GHGs)
- ✦ Calculation and data of CO₂ emissions
- ✦ Total GHG emissions of fuels and food

Evolution of air temperature at earth's surface for 1000 years



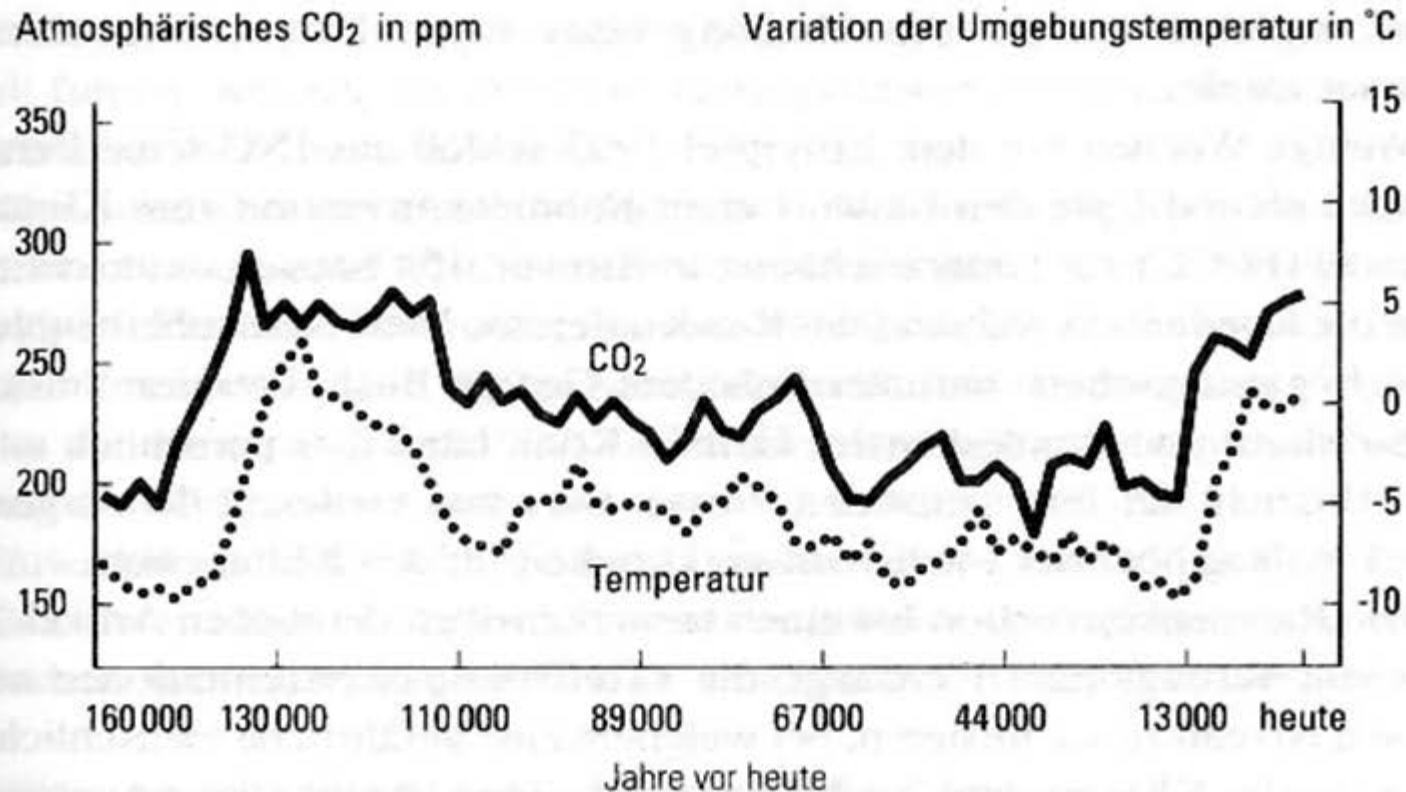
Source: renac

Evolution of CO₂ content in the atmosphere for 1000 years



Vostok Sensation:

Correlation between CO₂ and air temperature



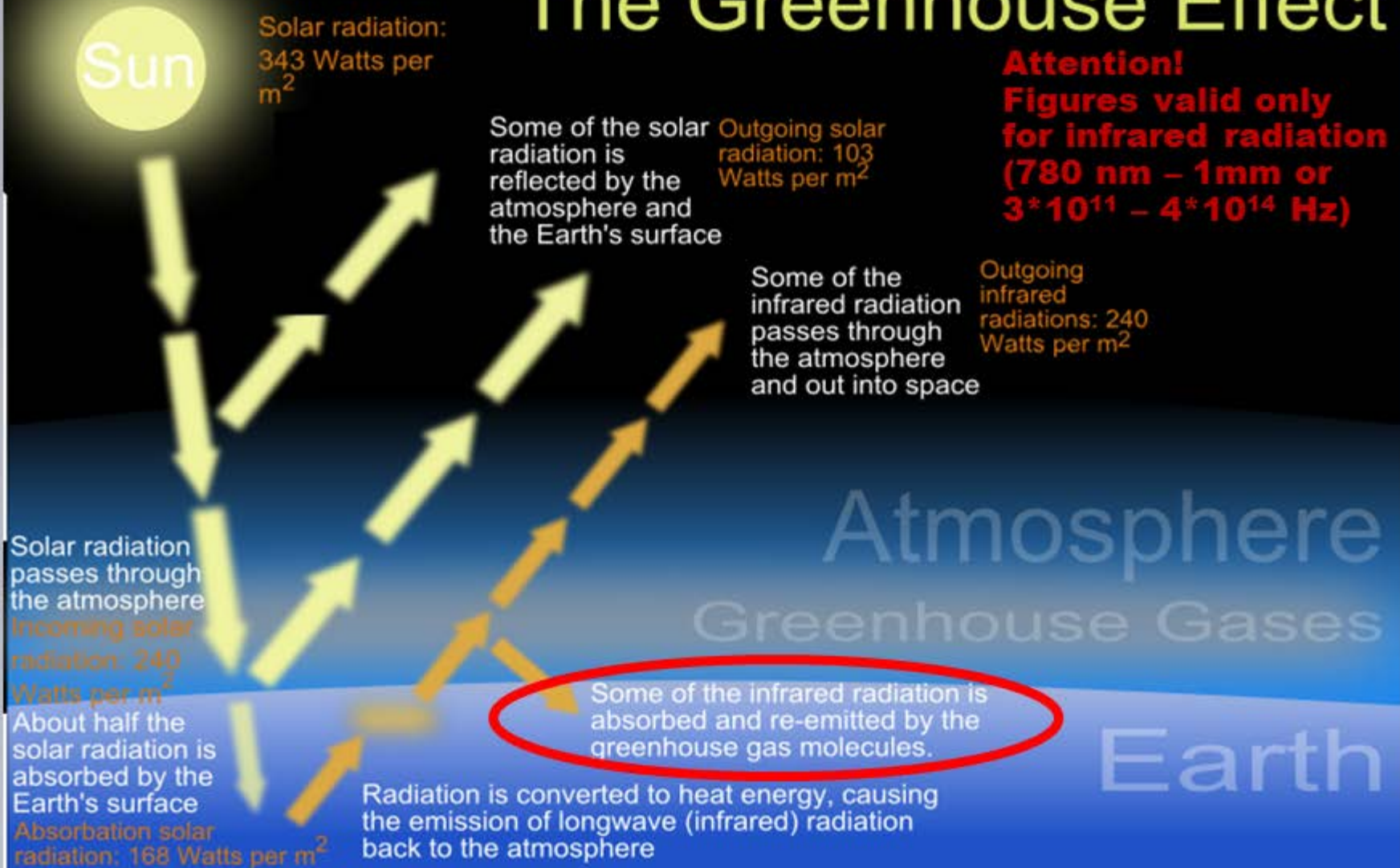
A strict correlation between CO₂ concentration and air temperature during the last 160000 years
measured in „fossile“ air bubbles in antarctic ice

Source: E. U. von Weizsäcker et al.: Faktor vier. München 1997

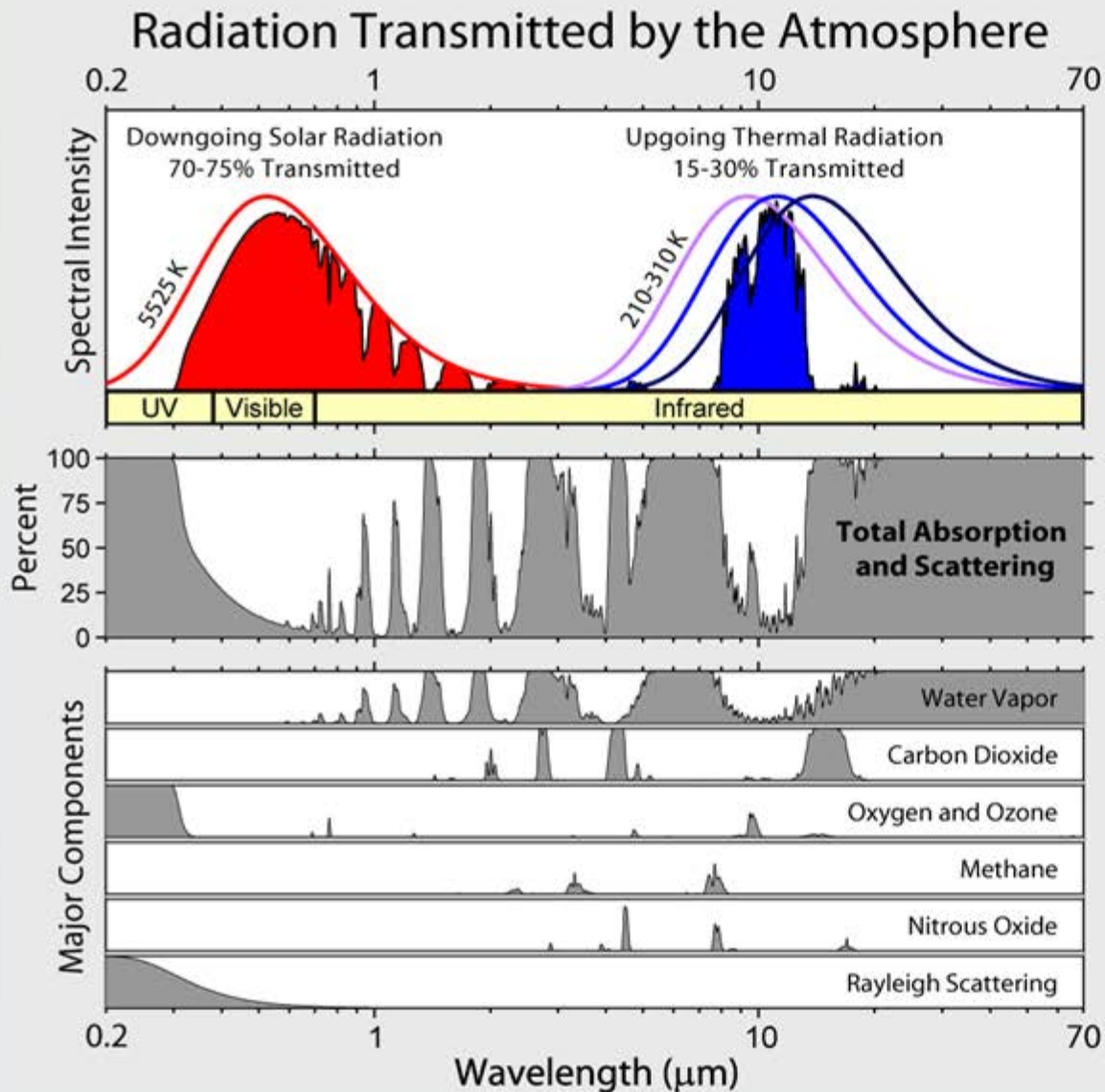
Greenhouse effect - The reason of global warming



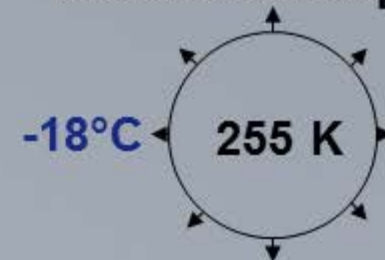
The Greenhouse Effect



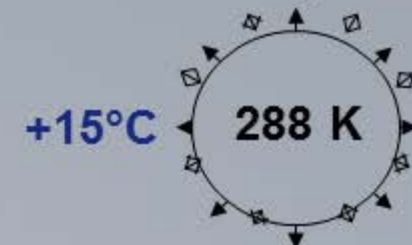
Physical reasons of the greenhouse effect



**Surface temperature
without atmosphere**



with atmosphere



Atmospheric absorption and scattering at different electromagnetic wavelengths. The largest absorption band of carbon dioxide is in the infrared.

Source: http://en.wikipedia.org/wiki/Greenhouse_gas

Greenhouse effect due to greenhouse gases (GHG)



Natural greenhouse effect

ca. 33 K (-18°C → +15°C)

Gas	Temperature contribution
H ₂ O	20.6 K
CO ₂	7.2 K
O ₃	2.4 K
N ₂ O	1.4 K
CH ₄	0.8 K
Other gases	0.6 K

Gas	Formula	Contribution in %
Water Vapor	H ₂ O	36 – 72 %
Carbon Dioxide	CO ₂	9 – 26 %
Methane	CH ₄	4 – 9 %
Ozone	O ₃	3 – 7 %

http://en.wikipedia.org/wiki/Greenhouse_gas

Anthropogenic greenhouse effect

ca. 0,5...1,0 K

Gas	contribution in %
CO ₂	50 %
CFC	22 %
CH ₄	13 %
O ₃	7 %
N ₂ O	5 %
H ₂ O (stratospheric)	3%

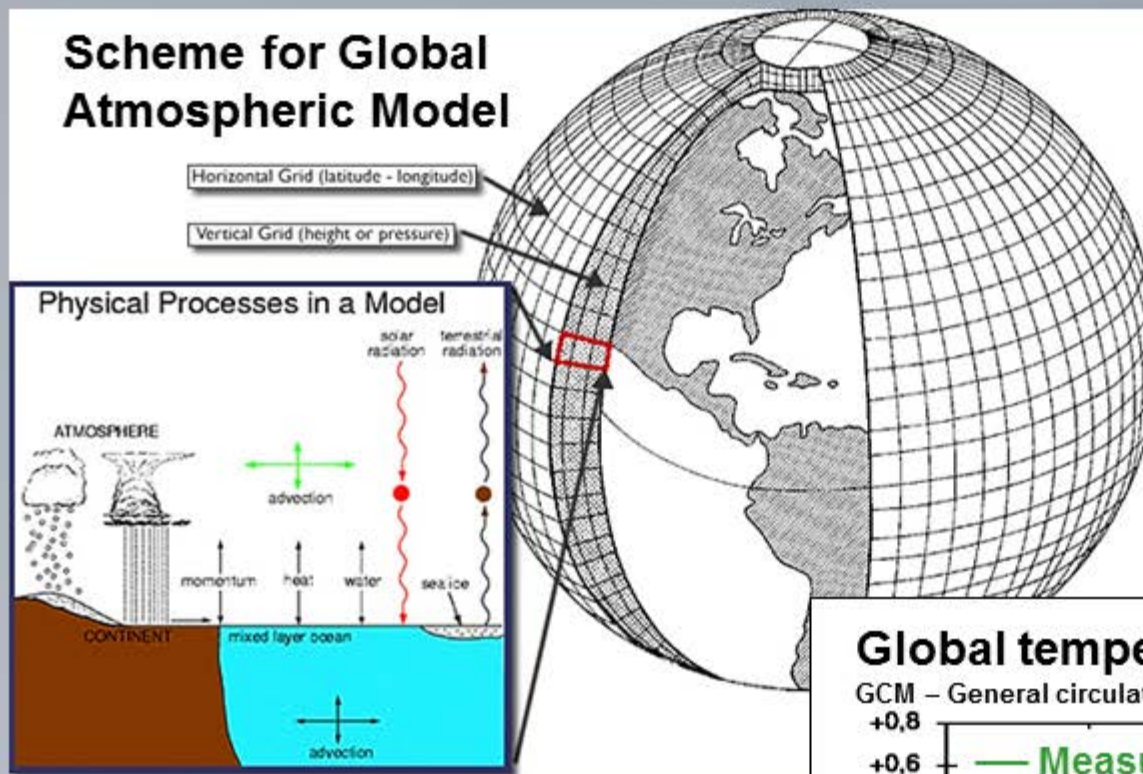
**See
also
later!**



CFC = Chlorofluorocarbons
N₂O = Nitrous Oxide

Prediction of climate change by global climate models

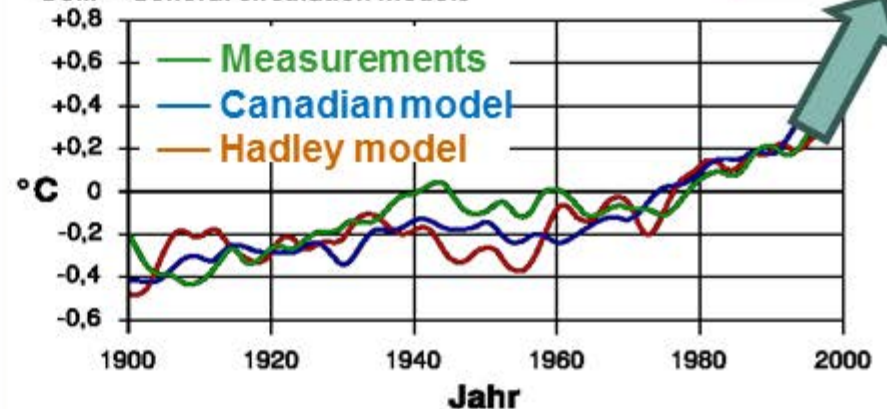
Scheme for Global Atmospheric Model



Climate models are systems of differential equations based on the basic laws of physics, fluid motion, and chemistry. To “run” a model, scientists divide the planet into a 3-dimensional grid, apply the basic equations, and evaluate the results.

Global temperature models

GCM – General circulation models



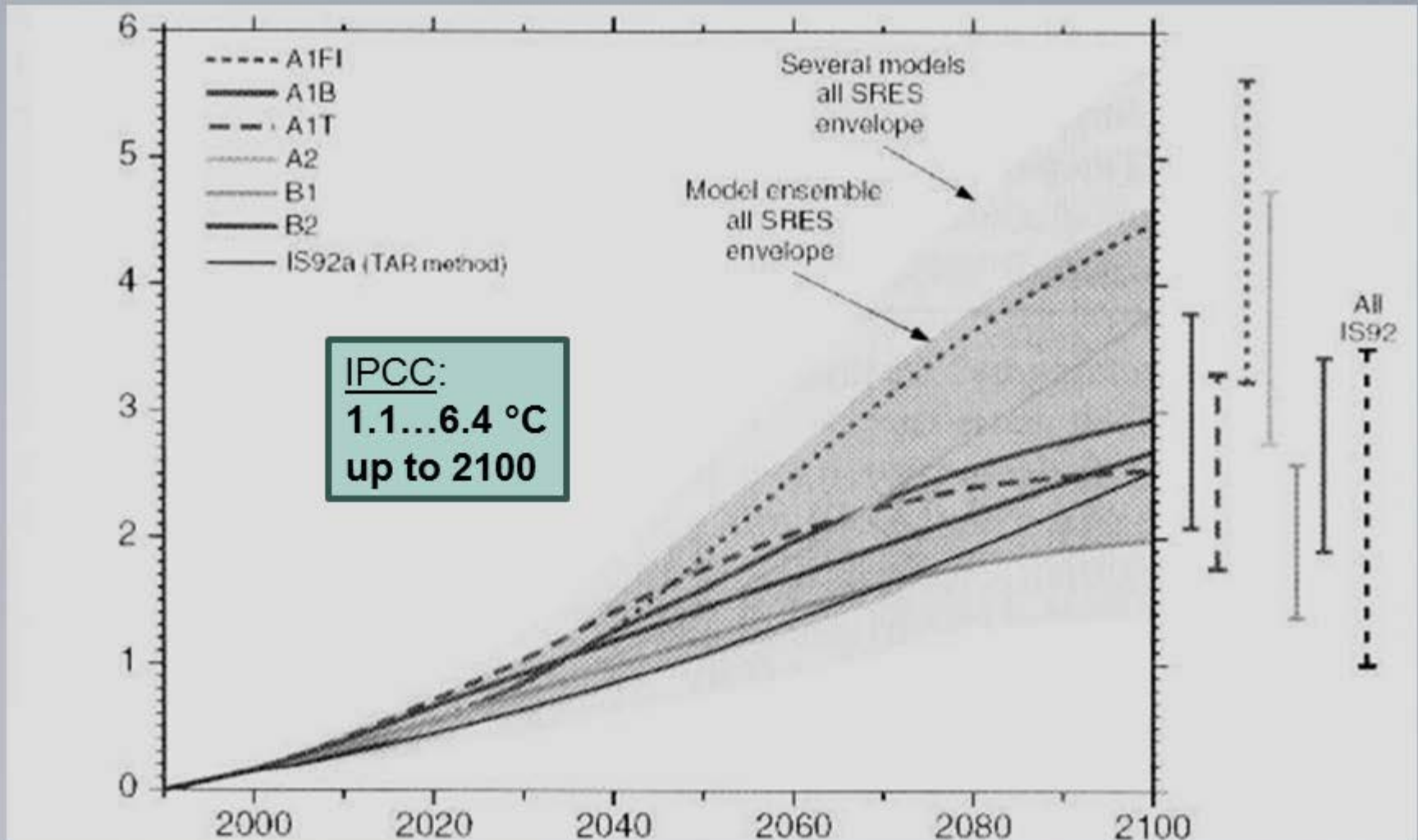
Future
see next!

Source: <http://de.wikipedia.org/wiki/Klimamodell>

Atmospheric models calculate winds, heat transfer, radiation, relative humidity, and surface hydrology within each grid and evaluate interactions with neighboring points.

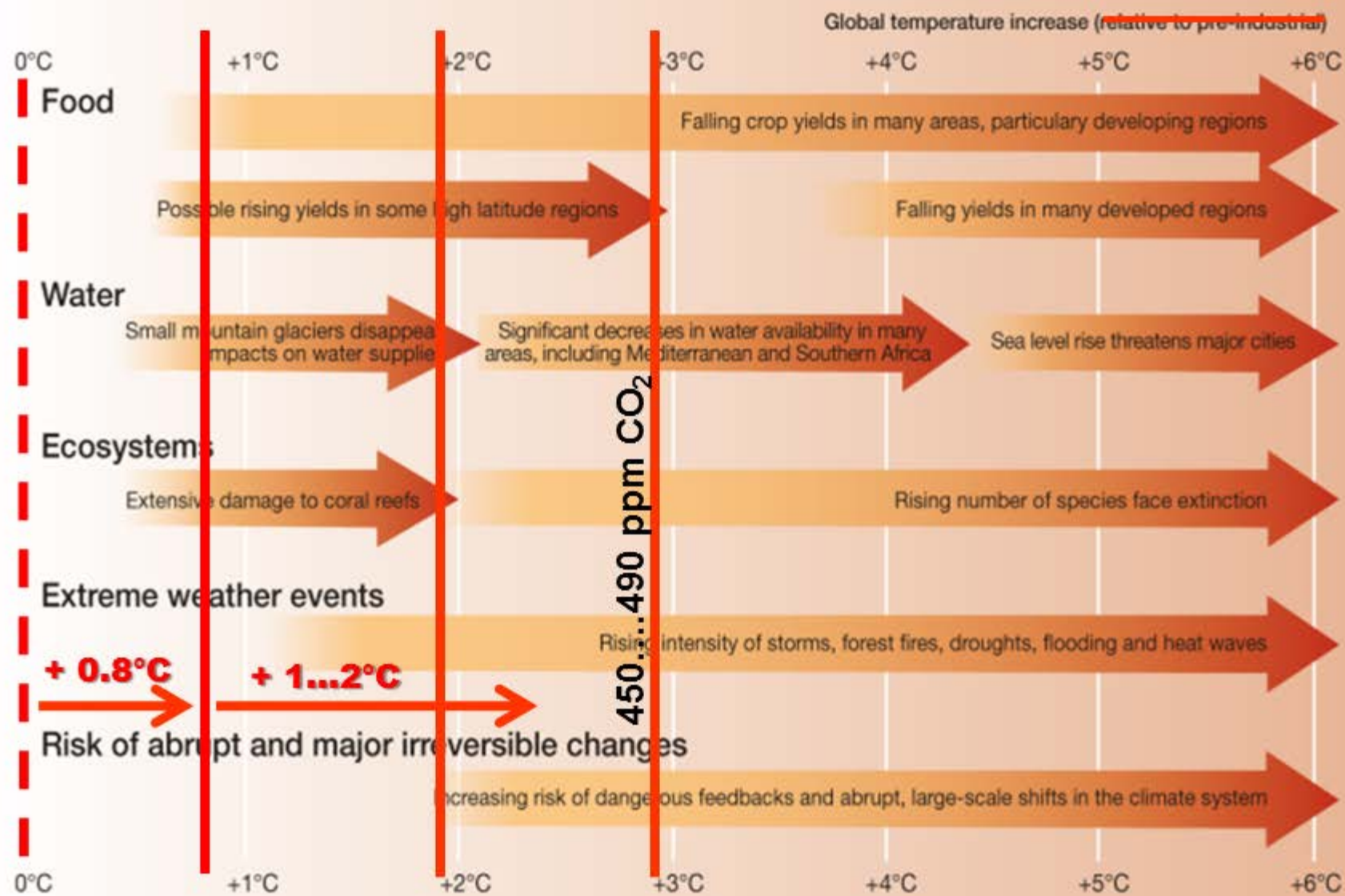
Temperature increase due to greenhouse effect

Selected forecast models up to 2100



Source: IPCC: Online report of the Intergovernmental Panel of Climate Change.

Impacts of climate change on human being and environment

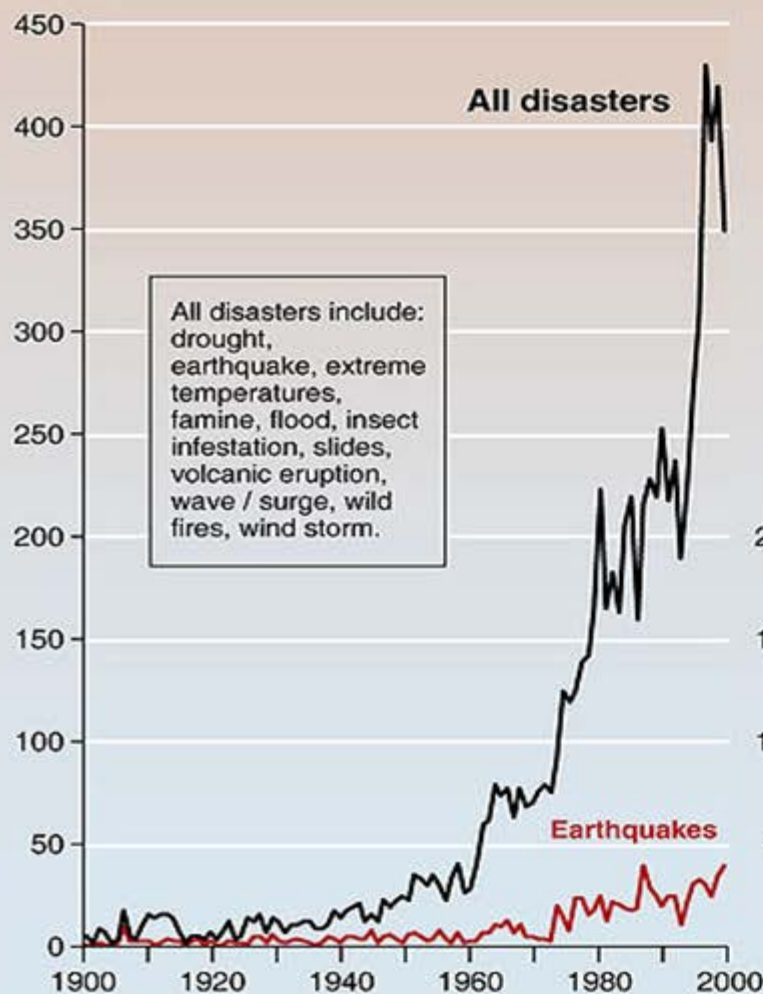


Source: renac

Increase in natural disasters

An effect of climate change?

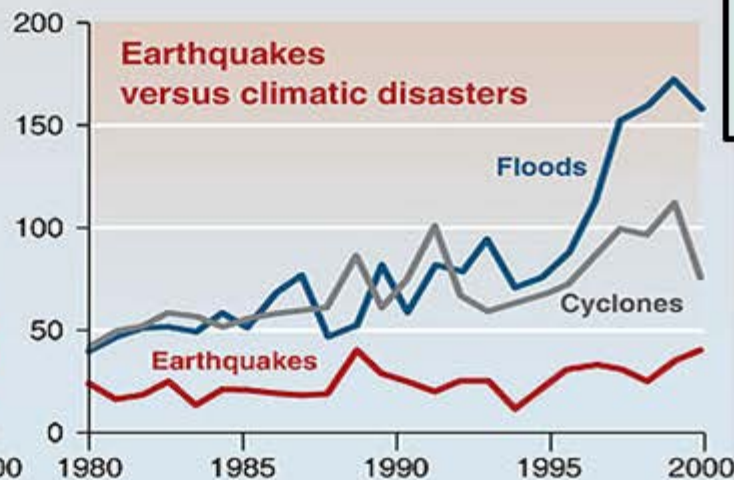
Number of events
per year



Source:

Trends in number of reported events

Much of the increase in the number of hazardous events reported is probably due to significant improvements in information access and also to population growth, but the number of floods and cyclones being reported is still rising compared to earthquakes. How, we must ask, is global warming affecting the frequency of natural hazards?



Reduction of flood and associated extreme weather costs is the primary benefit of climate change mitigation.

Source:
http://en.wikipedia.org/wiki/Mitigation_of_global_warming

Relevant GHGs and other pollutant gases



Climate gases (GHG)

(Kyoto gases)

CO ₂	-	Carbon dioxide
CH ₄	-	Methane
N ₂ O	-	Nitrous Oxide
SF ₆	-	Sulfur hexafluoride
PFC	-	Perfluorocarbons
HFC	-	Hydrofluorocarbons (HCF≈CFC=FCKW)

Source: Intergovernmental Panel Climate Change (IPCC)

Other pollutant gases

SO ₂	-	Sulfur dioxide
NO _x	-	Nitrogen oxide
HCl	-	Hydrochloric acid
HF	-	Hydrofluoric acid
CO	-	Carbon Monoxide
NMVOC	-	Volatile Organic Compounds hydrocarbons
H ₂ S	-	Hydrogen sulfide
NH ₃	-	Ammonia

Potential of the greenhouse effect

Equivalent of CO₂ (GWP)

CO ₂	Global Warming Potential (GWP) calculated for 100 years	1
CH ₄		21
N ₂ O		310
SF ₆		23.900
PFC		6.500 - 9.200
HFC		140 - 11.700

C₄H₁₀ (Butane)

4

Potential of acidification

Equivalent of SO₂

SO ₂	1
NO _x	0,696
HF	1,601
HCl	0,878
H ₂ S	0,983
NH ₃	3,762

Source: renac

Sources and impacts of the Kyoto gases



on global warming effect

Greenhouse gas (Kyoto gas)		Anthropogenic sources	Global warming potential	Anthropogenic share of global warming *
Carbon dioxide	CO ₂	Fossile energy sources	1	64%
Methane	CH ₄	Rice production, Cattle, Biomass	21	20%
Nitrous oxide	N ₂ O	Fertilizer, Fossile energy	310	6%
Hydroflourcarbons	HCFC	Cooling fluids	100-12000	10%
Perflourcarbons	PFC	Solvents, Alumin., Fire extenguisher	6500-9200	
Sulphur hexaflouride	SF ₆	Electrical indutry, Isolations	23900	

* <http://de.wikipedia.org/wiki/Emissionsrechtehandel>

Calculation of CO₂ emissions caused by combustion of fossil fuels



Oxidation of C to CO₂ (combustion of coal): $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ with
and oxidation C_xH_y to CO₂ and H₂O (combustion of oil and gas)

$$\Delta H = -393 \text{ kJ/mol}$$

Example Lignite:

1 kg coal = 500 g H₂O + 50 g ash + 135 g volatiles + 315 g C

315 g C → 1.155 kg CO₂ (with atomic weight C = 12 e O = 16)

1.155 kg / 44.0 kg/kmol (molecular weight) = 26.25 mol CO₂

To heat up 500 g H₂O from 10 to 100°C: - 0.188 MJ
and to evaporate 500 g H₂O (2.256 MJ/(kg K): - 1.128 MJ

$$\} \rightarrow 9.0 \text{ MJ/kg} = 2.5 \text{ kWh/kg}_{\text{coal}}$$

1.155 kg CO₂ / 2.5 kWh = 462 g CO₂/kWh

The oxidation of volatile produces more 3 MJ = 0.833 kWh → 345 g CO₂/kWh in total

→ A power plant with efficiency of 35 % produces approx. 1 kg CO₂/kWh_{el}.

Volume of mole:
CO₂ = 22.86 m³/kmol
Air = 22.47 m³/kmol

The exact figures

may vary because of varying compositions of fuels.

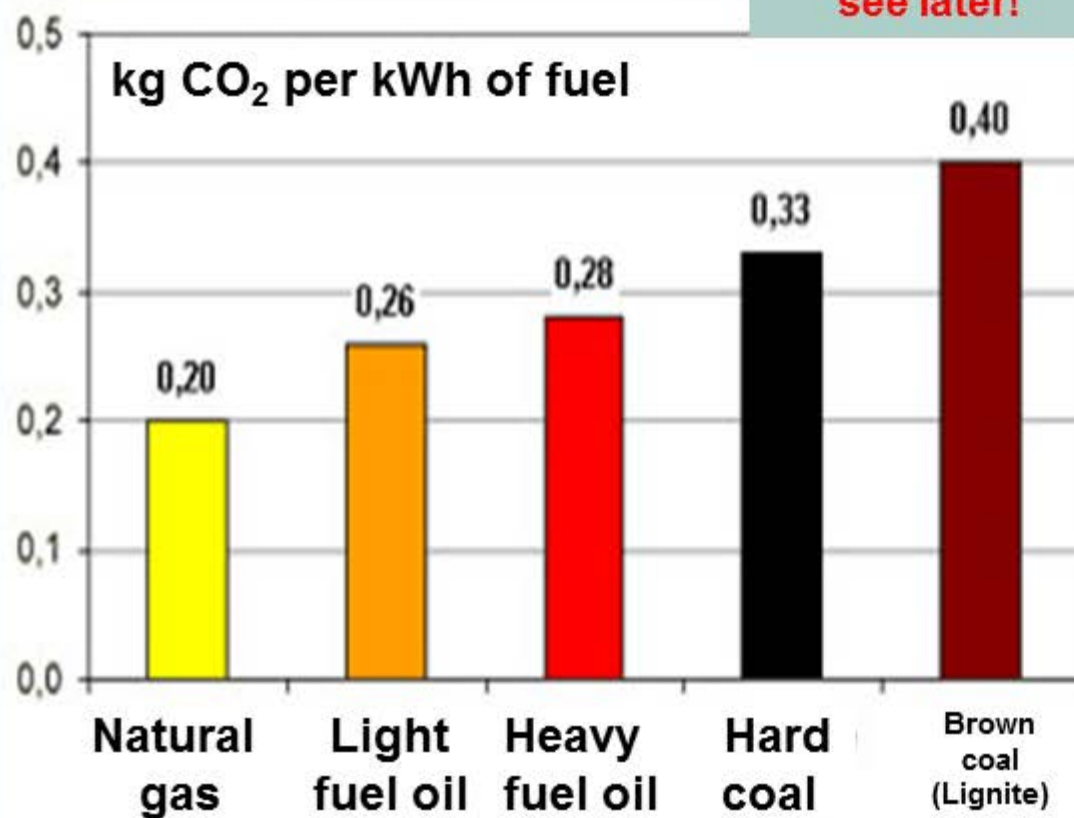
See next slides!

**Only CO₂
from combustion!**

Pure CO₂ emissions of the various types of fossil fuels



The total
GHG emissions
see later!



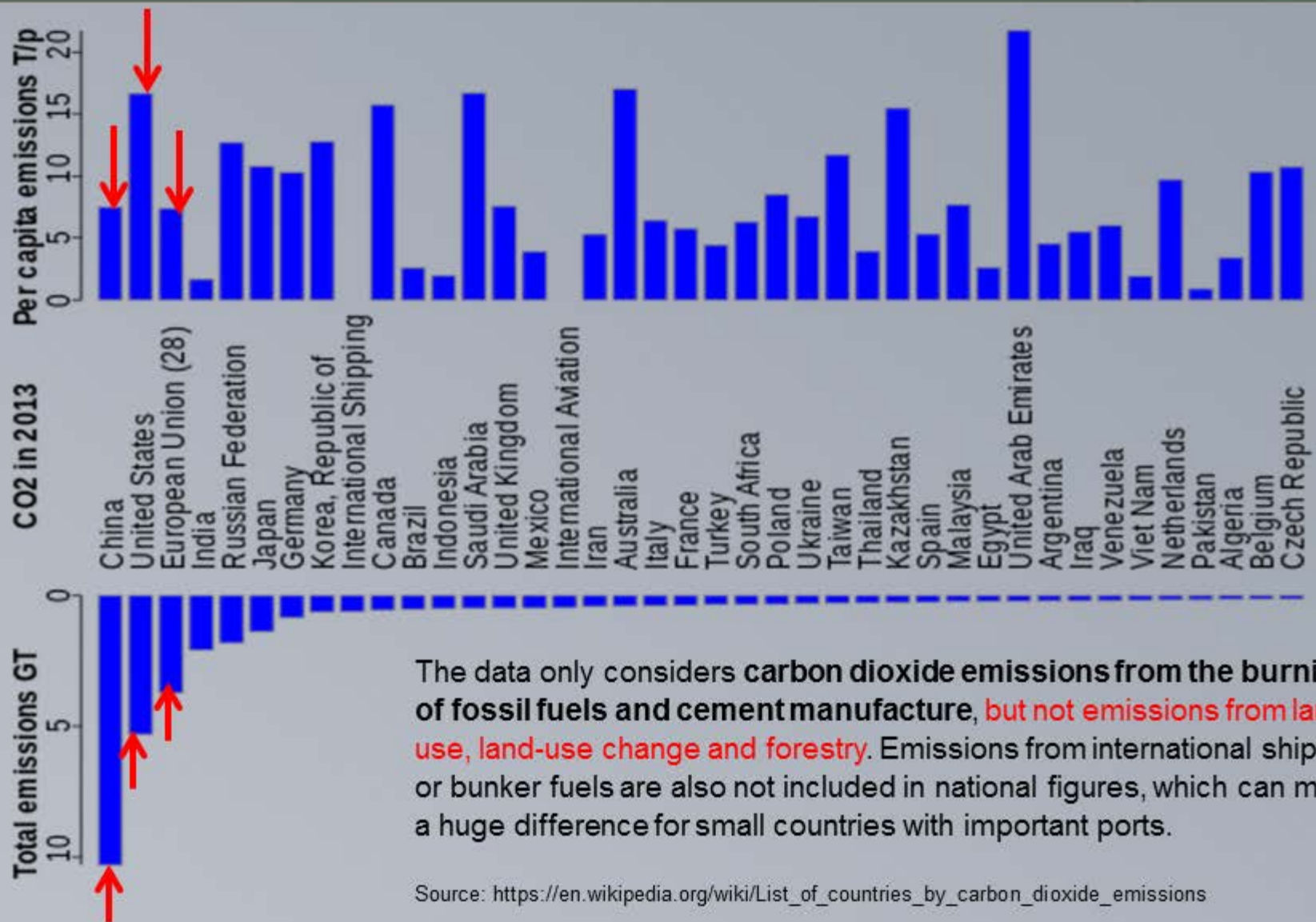
Source: http://www.baunetz.de/sixcms_4/sixcms/detail.php?id=142928

See also: IPCC emission factors

<http://www.carbonmetrics.com/ipcc.html>

Fuel name	CO ₂ emitted (lbs/10 ⁶ Btu)	CO ₂ emitted (g/10 ⁶ J)
Natural gas	117	50.30
Liquefied petroleum gas	139	59.76
Propane	139	59.76
Aviation gasoline	153	65.78
Automobile gasoline	156	67.07
Kerosene	159	68.36
Fuel oil	161	69.22
Tires/tire derived fuel	189	81.26
Wood and wood waste	195	83.83
Coal (bituminous)	205	88.13
Coal (subbituminous)	213	91.57
Coal (lignite)	215	92.43
Petroleum coke	225	96.73
Coal (anthracite)	227	97.59

Global CO2 emissions caused by fossil fuels



Total GHG emissions of fossil fuels



Total GHG emissions depend on fuel and utilization!

Fuel	Emission factors* for		
	Thermal use g CO _{2-equ.} /MJ	Transport g CO _{2-equ.} /MJ	Electricity g CO _{2-equ.} /kWh _{el}
Hard coal	88 - 106	-	822 - 1080
Brown coal	94 - 124	-	729 - 1230
Fuel oil	73 - 84	-	890 - 893
Diesel	-	84 - 91	-
Gasoline	-	84 - 94	-
Natural gas	51 - 68		(49) 599-751
Wood chips	10	-	-
Solar collector	10	-	-

* incl. the main GHGs and all processes for recovery, processing and transport of these sources

1 kWh = 3.6 MJ

Sources: <http://www.umweltdaten.de/publikationen/fpdf-l/3761.pdf>, http://www.bundestag.de/bic/analysen/2007/CO2-Bilanzen_verschiedener_Energietraeger_im_Vergleich.pdf
http://en.wikipedia.org/wiki/Emission_factor and Holzenergie-Fibel, Stuttgart 2003 and other

Total GHG emissions of power generation in Germany



No savings!

Electrical power from	CO ₂ equivalents* in g/kWh _{el}
Nuclear power station	16 - 126
Hard coal power station	622 - 1080
Brown coal power station	729 - 1230
Mineral oil	890
Natural gas power station or CHP	49 - 640
Wind power station	9 - 24
Hydroelectric power station	4 - 40
Photovoltaik plant	27 - 160
Biogas CHP plant	-400 - 170
<u>German mix (2012)</u>	<u>559</u>

**CO₂ savings e.g.
559 - (9-24) = 550-535**

* incl. all prior processes and material consumption for production of these plants

Source: http://www.bundestag.de/bic/analysen/2007/CO2-Bilanzen_verschiedener_Energietraeger_im_Vergleich.pdf

Total GHG emissions of liquid biofuels in g CO₂/kWh



Fuel (underlying raw material)	Emission factor [g CO ₂ -eq/kWh]
Petrol/diesel (fossil)	301.7
Biodiesel (rapeseed)	165.6
Biodiesel (soy)	180.0
Biodiesel (palm oil)	115.2
Biodiesel (waste)	36.0
Vegetable oil (rapeseed)	126.0
Vegetable oil (soy)	152.6
Bioethanol (grain)	180.0
Bioethanol (beets)	118.8
Bioethanol (sugar cane)	86.4
Bioethanol (other)	36.0
Biodiesel (weighted)	157
Vegetable oil (weighted)	126
Bioethanol (weighted)	132

**Average figures for Germany
without direct and indirect
land use changes**

**The GHG emissions of biofuels
result from:**

- use of fossil energy (CED) for production, conversion and transport of crops and biofuels
- the use of emissions causing materials
- emissions during cultivation of crops (N₂O, CH₄)
-

**but not from CO₂ emissions
during combustion of these fuels!**

Source: Renewable Energy in Figures. BMU 2010
http://www.bmu.de/english/renewable_energy/downloads/doc/5996.php
<http://www.umweltdaten.de/publikationen/fpdf-l/3761.pdf>

GHG savings of liquid biofuels in t CO₂/ha p.a.



Remember!

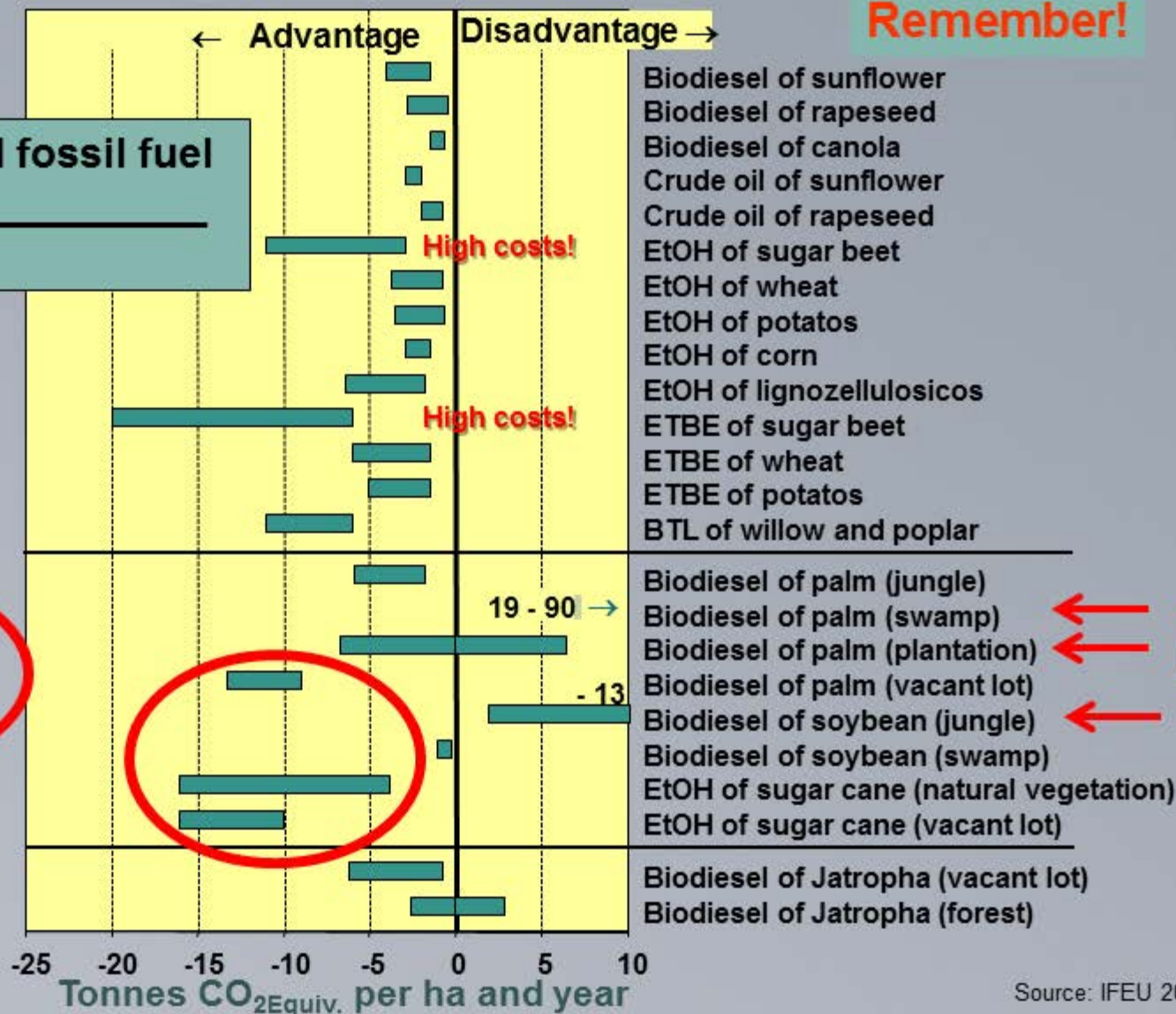
GHGs of substituted fossil fuel
- GHGs of biofuel

GHG savings

Biofuels from
temperate
areas

Biofuels from
humid tropical
areas

Biofuels from
arid areas

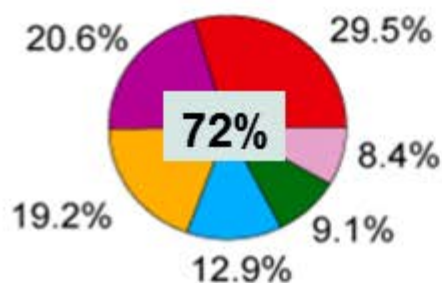
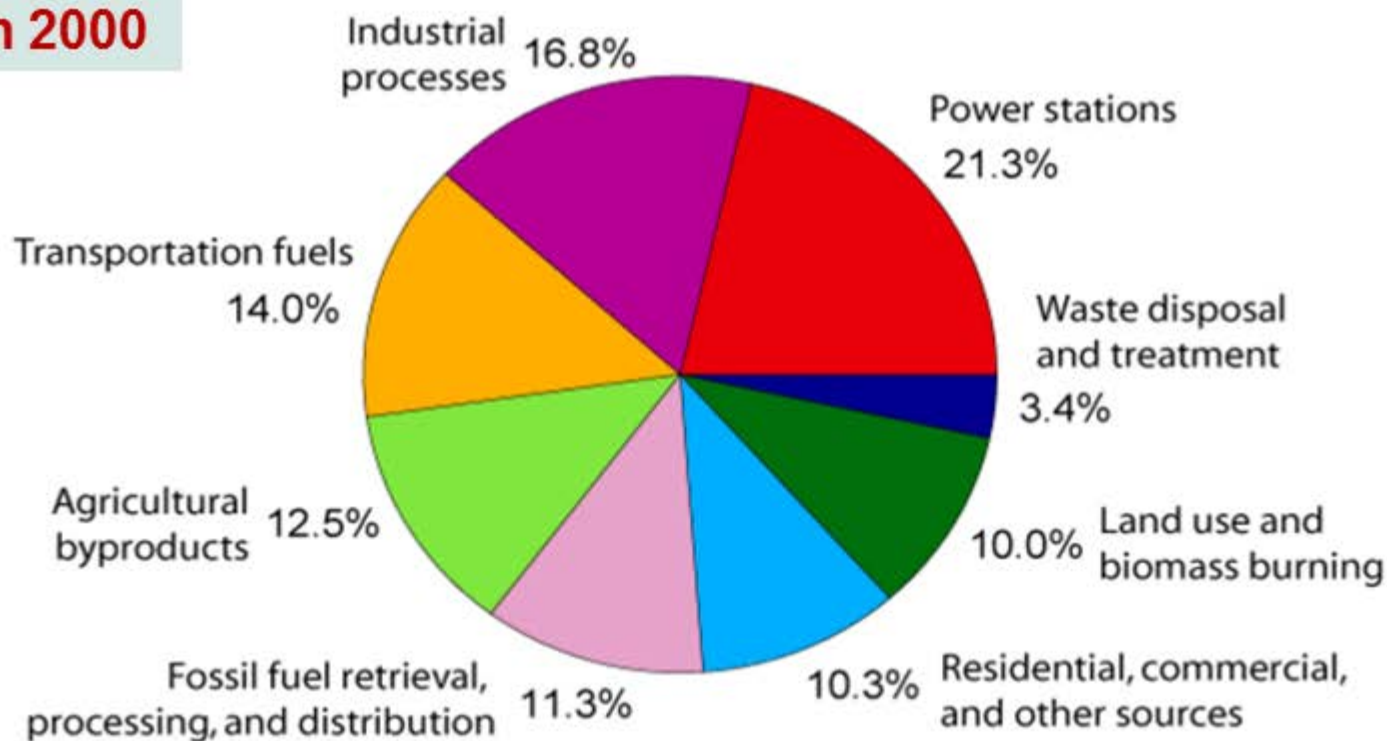


Source: IFEU 2008

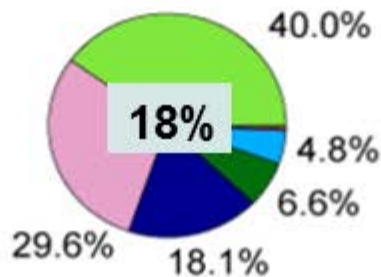
Global total GHG emissions by sectors incl. agriculture



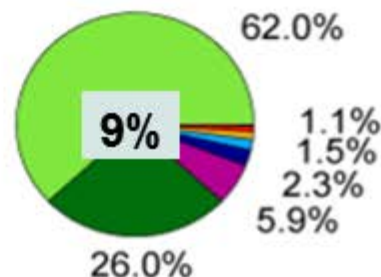
in 2000



Carbon Dioxide














Methane



Nitrous Oxide

Source: http://en.wikipedia.org/wiki/File:Greenhouse_Gas_by_Sector.png

GHG emissions of selected foods

	Butter	23500		Potato chips, frozen	5670
	Beef (meat)	13300		Bread	720
	Poultry (meat)	3490		Tomatos	310
	Pork (meat)	3250		Potatos	200
	Eggs	2570		Rice	4130
	Milk	940	CO ₂ equivalents in g CO ₂ /kg		

Source: Tagesschau.de In: <http://www.agenda21-treffpunkt.de/daten/treibhausgase.htm>

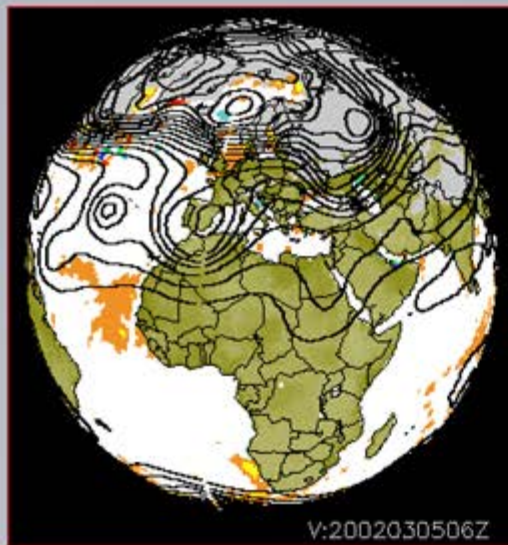
Calculation of GHG emissions by means of professional software



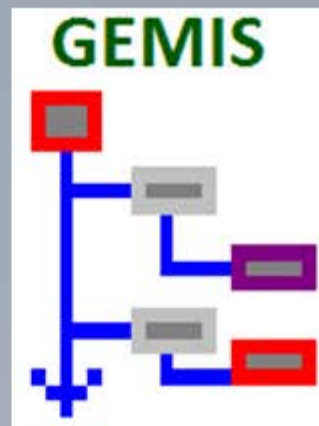
RETScreen® Canada

<http://www.nrcan.gc.ca/energy/software-tools/7465>

RETScreen is a Clean Energy Management Software system for energy efficiency, renewable energy and cogeneration project feasibility analysis as well as ongoing energy performance analysis.



Source: Environment Canada



Germany

www.iinas.org/gemis.html

IINAS is the host of **GEMIS** (Global Emissions Model for Integrated Systems), a public domain life-cycle and material flow analysis model and database that IINAS provides freely.

GEMIS was first released in 1989, and is continuously updated and extended since then. It is used by many parties in more than 30 countries for environmental, cost and employment analyses of energy, materials and transport systems.

**Freely available tools
to calculate your
ecological footprint**



Thank you

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